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Tandem Thien- and Benzannulations of α -Alkenoyl- α -alkynyl Ketene Dithioacetals with Cyanoacetates: Synthesis of Functionalized Benzo[b]thiophenes

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S Supporting Information

[AB](#page-2-0)STRACT: [A novel dom](#page-2-0)ino annulation strategy for the construction of benzo $[b]$ thiophenes has been developed. In the presence of Cs_2CO_3 and Ag_2CO_3 , a wide range of α alkenoyl- α -alkynyl ketene dithioacetals readily react with cyanoacetates in CH₃CN at 110 °C under N_2 to afford multisubstituted benzo $[b]$ thiophenes efficiently via tandem thien- and benzannulations. A plausible mechanism is also proposed.

 \mathbf{B} enzo[b]thiophenes represent a class of important
heterocyclic compounds which are widely distributed in bioactive natural products.¹ These heterocycles form the core of a number of medicinally important molecules, such as raloxifene,^{2a} arzoxifene,^{2b} [zi](#page-2-0)leuton,^{2c} sertaconazole,^{2d} and SB-271046^{2e} (Figure 1). Furthermore, benzo[b]thiophenes are also wide[spr](#page-2-0)ead in cata[lys](#page-2-0)is and m[at](#page-2-0)erial chemistry.

Figure 1. Pharmaceuticals containing the benzo $[b]$ thiophene core structure.

Two synthetic strategies for benzo $[b]$ thiophenes are commonly used.⁴ The first strategy is the thienannulation⁵ onto a benzene precursor (Scheme 1a).6,7 Typical examples include the elect[ro](#page-2-0)philic cyclization of o-alkynyl thioanisoles, [as](#page-2-0) explored by Larock.⁶ Recently, König [et](#page-2-0) al.^{7b} reported a photocatalytic reaction of o-(methylthio)arenediazonium salts with alkyn[e](#page-2-0)s for the synthesis of benzo $[b]$ t[hio](#page-2-0)phene. The second strategy is the benzannulation on a preformed thiophene moiety (Scheme 1b).⁸ One elegant example is the gold-catalyzed benzannulation of 2-substituted thiophene, reported by Hashmi and co-[wo](#page-2-0)rkers.^{8b} A strategy toward synthesis of benzo $[b]$ thiophenes via tandem thien- and benzannulations of simple acyclic precu[rso](#page-2-0)rs is an appealing alternative but has not been developed. To the best of our

Scheme 1. Different Strategies for the Construction of Benzo[b]thiophenes

Ag₂CO₂, Cs₂CO₂ COOEt CH₃CN, N₂, 110 °C 24 examples up to $92%$

a) Thien-annulation of benzene derivatives

b) Benz-annulation of thiophene derivatives

$$
\begin{array}{ccc}\n\mathbb{R}^R & \longrightarrow & R^T \longrightarrow & S^T \\
\mathbb{R}^R & \longrightarrow & R^T \longrightarrow & S^T\n\end{array}
$$

c) This work: Thien- and benz-annulation of ketene dithioacetals 1

knowledge, up to now, there has been only one single example reported by Mukai wherein pyrolysis of bicyclic $δ$ -thia- $α, β$ unsaturated ketone at 520 $^{\circ}$ C afforded 4-hydroxybenzo[b]thiophene in 28% yield.⁹ Although it remains unexploited, this strategy would be highly desirable because it would avoid using prefunctional aromatic ring substrates and the functional groups could be directly introduced onto the benzo[b] thiophene skeletons by choosing appropriate substrates.

Tandem reactions have attracted considerable attention for simplifying reaction steps, reducing waste, and maximizing atom economy and have been developed as a powerful shortcut for the assembly of ring systems in organic synthesis.¹⁰ Among these processes, the cyclization of enynes is a rapid and powerful approach to prepare cyclic derivatives.^{10,11} [He](#page-3-0)rein, we report new tandem cyclization reactions of α-alkenoyl-α-alkynyl ketene dithioacetals $1^{12,13}$ with ethyl 2-c[yano](#page-3-0)acetate to synthesize benzo $[b]$ thiophene derivatives (Scheme 1c). To

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the best of our knowledge, this is the first efficient approach for the synthesis of benzo $[b]$ thiophenes by the construction of both benzene and thiophene rings in one pot. The readily availability of starting materials enables efficient access to highly functionalized benzo $[b]$ thiophenes that might find applications in the pharmaceutical industry and material science.

We commenced our studies by exploring the reaction between (E)-4-(1,3-dithiolan-2-ylidene)-1,6-diphenylhex-1-en-5-yn-3-one (1a) with ethyl 2-cyanoacetate to optimize the reaction conditions (Table 1). When 1a was treated with 4.0

OH \ll ^{Ph} Ag_2CO_3 base NC. COOEt Ph Ph $+$ solvent, N ₂ Ph ⁻ Phi Phi temp, 5h ĊΝ ĊΝ 3a 2a 1a					
				yield b (%)	
entry	base	solvent	temp $(^{\circ}C)$	2a	3a
1^c	DBU	CH ₃ CN	90	15	35
\overline{c}	DBU	CH ₃ CN	90	50	9
3	K_3PO_4	CH ₃ CN	90	40	15
$\overline{4}$	K_2CO_3	CH ₃ CN	90	31	12
5	NaOH	CH ₃ CN	90	37	20
6	Cs_2CO_3	CH ₃ CN	90	62	10
7	Cs ₂ $CO3$	CH ₃ CN	110	88	trace
8	Cs_2CO_3	CH ₃ CN	120	69	trace
9	Cs ₂ $CO3$	DMSO	110	50	10
10	Cs , $CO3$	toluene	110	θ	Ω
11		CH ₃ CN	110	θ	trace
12^d	Cs_2CO_3	CH ₃ CN	110	trace	73

a Conditions: 1a (0.2 mmol), ethyl 2-cyanoacetate (4.0 equiv), Ag_2CO_3 (1.0 equiv), base (1.5 equiv), and solvent (2.0 mL) for 5 h $\mu_{\rm B2}$ isolated yields. ^c10 mol % Ag₂CO₃ was used. ^dThe reaction under N₂. ^bIsolated yields. ^c10 mol % Ag₂CO₃ was used. ^dThe reaction was run without Ag_2CO_3 .

equiv of ethyl 2-cyanoacetate, 10 mol % Ag_2CO_3 , and 0.5 equiv of DBU in CH₃CN under N₂ at 90 °C, the desired product $2a^{14}$ was obtained in 15% yield, with 3a (35%) being the major product (entry 1). As expected, increasing the loading [of](#page-3-0) Ag_2CO_3 to 1.0 equiv improved the yield of 2a to 50% (entry 2). Changing the oxidants (i.e., AgTFA, O_2 , AgN O_3) did not afford 2a or led to reduced yields (see details in the Supporting Information). We then turned our attention to screen different bases (entries 3–6). The results indicated that $Cs₂CO₃$ could [remarkably](#page-2-0) improve the yield to 62% (entry 6). [Among](#page-2-0) [the](#page-2-0) reaction temperatures examined, it turned out that the reaction at 110 °C gave the best result (entries 6–8). CH₃CN was found to be an optimal solvent in comparison with DMSO and toluene (entries 9 and 10). Control experiments verified the requirement of Cs_2CO_3 and Ag_2CO_3 (entries 11 and 12).

With the optimized conditions in hand, the substrate scope was then investigated. Overall, a number of multisubstituted benzo[b]thiophenes were successfully prepared with moderate to excellent yields (Scheme 2). First, substituents on the alkenoyl moiety were investigated (2a−l). Arylalkenoyl substrates 1 with electron-donating groups showed better reactivity than those with electron-withdrawing groups. Various kinds of functional groups, such as OMe, $NO₂$, F, Cl, and Me, were well tolerated. Naphthyl, heteroaryl, and ferrocenyl substrates were also productive, delivering benzo $[b]$ thiophenes in moderate yields (2i−l). Furthermore, substituents on the terminal alkyne were investigated. A variety of electron-

^aReaction conditions: 1 (0.2 mmol), ethyl 2-cyanoacetate (4.0 equiv), Ag₂CO₃ (1.0 equiv), and Cs₂CO₃ (1.5 equiv) at 110 °C for 5 h under N_2 . b Isolated yields.

deficient and electron-rich arylalkynyl substrates underwent the annulation to afford benzo $[b]$ thiophenes in fair to excellent yields (2m−t). Efficiency was not much influenced by electronic variation on the aryl moiety at the terminal alkyne. Moderate yields were found to be achievable from heteroaryl alkyne and aliphatic alkyne substrates (2s and 2t). Notably, substrate 1x with TMS substituent on the alkyne underwent the annulation reaction to give the de-trimethylsilylation product 2x. However, substrate 1y with terminal alkyne could not be transformed into $2x$ under the current conditions (eq 1). These results indicate that de-trimethylsilylation proceeded post to the annulation of 1x.

To gain some preliminary understanding of the reaction mechanism, the following experiments were carried out (Scheme 3). Resubjection of 3a, which was found during early optimization experiments, to the standard conditions did result in t[he](#page-2-0) formation of the benzo $[b]$ thiophene 2a (Scheme 3a), suggesting the intermediacy of 3a in the reaction. Besides,

Scheme 3. Preliminary Mechanistic Studies

we investigated the reactivity of thiophene 4, which might be formed from 1a under base.^{13d} Indeed, upon reaction with ethyl 2-cyanoacetate, 4 was also converted to give benzo $[b]$ thiophene 2a in 76% yield (S[che](#page-3-0)me 3b). This result suggested that benzo $[b]$ thiophene construction might go through the thienannulation/benzannulation sequence in this reaction.

Although the precise reaction mechanism cannot be definitively established at the current stage, on the basis of the above results and previous reports, $13a,d$ a plausible mechanism was proposed in Scheme 4. Initially, deprotonation at one of the

Scheme 4. Putative Reaction Mechanism

methylene groups of the dithiolane moiety $13d,15$ triggered the ring-opening reaction to generate the thiolate anion II , 13d which cyclized to form thiophene interm[ediate](#page-3-0) 4 through a sequential intramolecular heteroannulation¹⁶ and protonati[on.](#page-3-0) Subsequently, Michael addition of the anion of ethyl 2 cyanoacetate to the double bond of 4 follo[we](#page-3-0)d by protonation formed intermediate IV.^{13a} Deprotonation of IV triggered an intramolecular nucleophilic displacement of the labile thioether furnishing VII,^{17} which [und](#page-3-0)erwent ester elimination 18 followed by oxidative aromatization¹⁹ to produce the desired benzo[b]thiophenes 2. [E](#page-3-0)xistence of intermediates IV and [V](#page-3-0)II were supported by HRMS anal[ysi](#page-3-0)s.²⁰

In summary, a novel tandem annulation of α -alkenoyl- α alkynyl ketene dithioacetals w[ith](#page-3-0) ethyl 2-cyanoacetate has been developed, yielding multifunctionalized benzo[b]thiophenes in moderate to excellent yields. This work opens up a new approach to realize benzo $[b]$ thiophene synthesis by tandem thien- and benzannulations in one pot. Ongoing studies are focused on applying this methodology in the synthesis of functional materials as well as testing bioactivity of some compounds.

■ ASSOCIATED CONTENT

S Supporting Information

Experimental procedures, analytical data for all compounds, and crystallographic data (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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